

Finally, pores were generated in the graphene membranes by oxidative etching on exposure to ultraviolet light. The molecular selectivity of these membranes was then demonstrated by the observation that δ decreases two orders of magnitude more rapidly as compared with nonporous graphene for small gas molecules such as H₂ and CO₂, while the rate

remains largely unchanged for Ar and gas molecules with a larger kinetic diameter.

In graphene membranes that have larger pores, faster leak rates were measured using a mechanical resonance method which again demonstrated sizeselective leak rates, but this time for molecules above and below the size of SF₆.

"Our results are consistent with theo-

retical models based on effusion through Ångstrom-sized pores," stated group leader J. Scott Bunch, "and represent an important step toward the realization of macroscopic, size-selective porous graphene barriers."

Rich Louie

Energy Focus

Energy harvesting in electronic displays enabled by fluorescentdye-cascade linear polarizer

s anyone with a smart phone or A tablet knows, battery life is the key to a useful mobile device. Recently, luminescent solar concentrators (LSCs) have gained attention as a possible energy-harvesting technique to extend mobile battery life. When integrated into a display, LSCs steer incoming light that would otherwise be wasted (sunlight, room lights, or the internal backlight) to small photovoltaics placed inside the device perimeter. However, current LSCs reduce display quality and therefore remain impractical. Now, A.M. Velázquez and colleagues at the Massachusetts Institute of Technology have demonstrated an LSC based on cascading fluorescence that may eliminate this problem by shifting reemitted light into the infrared. They reported their findings in the October 25, 2012 online edition of Energy & Environmental Science (DOI: 10.1039/c2ee23265k).

LSCs studied to date have been based on luminescent dye molecules placed

in a glass sandwich waveguide. These molecules can be aligned within a liquid crystal host, making the system into a linear polarizer, which could potentially replace the standard linear polarizers used in display technology. However, photons that result from the dye fluorescence can escape from the waveguide and be reemitted from the surface, contaminating the display image. The researchers hypothesized that they could overcome this problem by combining four different fluorescent dyes, with emission and absorption spectra that overlap to cover the optical spectrum all the way to the infrared. The final emission of this cascade could therefore be filtered out by a standard infrared filter.

The researchers prepared the device by first dissolving four dyes in chloroform, mixing them with a nematic liquid-crystal host at 60°C for five minutes and desiccating the mixture for 12 hours. Next, they used capillary action to fill a 5-µm-gap liquid-crystal cell with the mixture, heating the materials to 70°C and slow-cooling them to ensure good alignment on the rubbed polyamide surfaces of the cell. The dyes chosen were (in wt% in the final device) coumarin 6

(0.3%), DCM2 (0.3%), Nile Red (0.6%), and a squaraine dye (0.2%).

Using an integrating sphere to make optical measurements, the researchers found that the final device exhibits good optical quantum efficiency: approximately 20% of incident photons at the peak absorption wavelengths are steered to the perimeter, which is in good agreement with theory. This suggests that the cascade of multiple dyes undergoes Förster resonant energy transfer with approximately unity efficiency. The combination of dyes selected also ensured that the photoluminescence of the dye cascade peaks in the nearinfrared, meaning that light escaping the waveguide could be easily filtered without affecting display quality. If used in a standard crossed-polarizer display, the researchers estimate that the device could generate over 10 µW/cm² indoors and as much as 1-10 mW/cm² outdoors.

These results suggest that displays on mobile devices may someday do double duty, providing information for the user while also generating electrical power for the device.

Colin McCormick

Bio Focus

Bioinspired polymeric carbon nitride nanospheres dramatically improve photoredox catalysis

any attempts have been made to create synthetic analogues of the natural photosynthesis system. However, the ability to mimic the spatial organization of this system, where the thylakoid membrane acts as a photosynthetic scaffold that captures and distributes light, and separates the oxidative from reductive species, remains a significant challenge. As reported in the October 16, 2012 online edition of Nature Communications (DOI: 10.1038/ ncomms2152), X. Wang and co-researchers at Fuzhou University, China,

and M. Antonietti at the Max Planck Institute of Colloids and Interfaces, Potsdam, Germany, have addressed this challenge by employing polymerized hollow nanospheres as analogues of the thylakoid membrane of a chloroplast. These hollow carbon nitride nanospheres (HCNSs) were used as a light-harvesting platform for catalyzing hydrogen evolution under visible light