AN IMPROVED METHOD FOR RADIOCARBON DATING FOSSIL BONES

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ABSTRACT. We demonstrate here that the use of either the acid-alkali or Longin method alone does not fully extract contaminants from bone-collagen fractions. We present our methods and results of extracting bone collagen that involve successive use of both of these methods.

INTRODUCTION

Radiocarbon dating of fossil bones, including, for example, mammoth teeth and antler, is of great interest because such material is the only organic matter remaining in Quaternary sediments and Paleolithic sites. It is well known that the organic matter of bones, comprising 20% of body mass, consists mainly (80–90%) of very thin collagen fibers, closely connected with hydroxyapatite crystals, $3Ca_3(PO_4)_2 \cdot Ca(OH)_2 + carbonic acid salts, which represent most of the non-organic component of bone. Because the crystals are very small (<math>200 \times 400 \times (15-30)$ Å), the surface area of bone is rather large, $>100m^2$ g⁻¹ (Engström *et al.* 1958).

However, the large surface area and porous structure of bone make it an excellent medium for sorption and precipitation of dissolved and colloidal particles of humic material, which, in groundwater, derives from modern soils and lacustrine-bog deposits. Such mixing of modern with ancient deposits often reduces the ¹⁴C age of fossil bones. Saprophytes and bacteria that assimilate CO₂ from the air can be another source of bone contamination.

We discuss the problems of dating fossil bones using only organic components. We share the widespread, experimentally proven view that the carbonate fraction of bones, including the fraction connected with apatite, is unsuitable for ¹⁴C dating because of contamination by groundwater carbonates and atmospheric CO₂ resulting from adsorption, recrystallization and isotopic exchange (Hassan, Termine and Haynes 1977).

The acid-alkali method (HCl-NaOH) of collagen extraction and purification from humic contaminants was developed in the late 1960s (Berger and Libby 1966; Haynes 1967; Vogel and Waterbolk 1967; Arslanov and Gromova 1970). Longin's (1971) method of collagen extraction from pulverized bone by 8% HCl and subsequent purification by dissolution (pH = 3, T = 90°C) also became accepted.

METHODS AND RESULTS

The use of collagen for ¹⁴C dating, after humus removal by NaOH extraction or by Longin's method of humus precipitation, is based on the questionable assumption that all humic impurities are removed completely from the collagen. Because part of the humic matter in fossil bones might occur as compounds hardly related to the mineral component of the sample, or as insoluble

complexes with fulvic acids, some highly polymerized and nearly insoluble compounds might not be extracted from the collagen by NaOH treatment at room temperature. On the other hand, extracting collagen by Longin's method might not transfer all the humic matter to the residue; a less polymerized and more soluble portion could be dissolved in the hot, slightly acidic solution and transferred in solution together with gelatin. Thus, contamination of extracted substances by humic matter is possible in both cases. The reliability of these methods has been checked by dating young fossil bones using different fractions of organic matter and by parallel dating of collagen-charcoal and gelatin-charcoal pairs (Haynes 1967; Longin 1971).

Minute carbon contamination of relatively young fossil bones changes their age only slightly and is difficult to detect. However, contamination by young carbon strongly affects the ¹⁴C dates of Pleistocene fossil bones; collagen content decreases with age and humic impurities accumulate in the porous bone structure. To evaluate contamination by young carbon, we collected bone samples from Late Pleistocene geological deposits and Paleolithic sites.

We investigated the solubility of bone organic matter as a function of the concentration of HCl and NaOH solutions. This enabled us to determine the best conditions of collagen extraction by HCl treatment of bones and subsequent removal of humic contaminants by NaOH treatment. We used modern cow bones and fossil bones that contained humic impurities. We added pulverized (≤ 1 mm) sample splits to HCl solutions of different concentrations and let them stand at room temperature for 24 h. Collagen was separated by centrifugation and washed until it showed a negative reaction for Ca²⁺ ((NH₄)₂C₂O₄ solution test). We then determined the carbon content of the samples. Then, dried collagen (3 g) was treated with different concentrations of NaOH solution for 24 h at room temperature. The collagen was washed following removal from the alkali extract until it became neutral; then we dried it in vacuum and determined its carbon content. We also analyzed the acid and alkali solutions for organic carbon content. Table 1 shows the results.

TABLE 1. Organic Components of Bones Dissolved at Room Temperature for Different Concentrations of HCl and NaOH*

Sample	Concentration of HCl or NaOH (N)	Weight of insoluble collagen (g)	Quantity of carbon in collagen (g)	Weight of carbon in soluble fraction (g)	Part of organic carbon in soluble fraction (%)
		HCl			
Modern bone	0.25**	16.96	5.99	0.65	9.8
(50 g)	0.5	13.40	6.12	0.62	9.2
	1	11.20	5.90	0.78	11.7
	2	10.11	5.36	1.34	20.0
Old bone	0.5	17.40	4.65	0.63	11.9
	1	15.74	4.51	0.80	15.1
	2	14.41	4.11	1.16	22.0
		NaOH			
Modern bone	0.1	2.974	1.567	0.0824	3.45
(3 g collagen)	0.25	2.608	1.374	0.1742	11.25
	0.5	2.475	1.304	0.2414	15.62
	1.0	2.122	2.118	0.6724	37.56
Old bone	0.1	2.7746	0.796	0.1225	13.33
(3 g collagen)	0.25	2.5070	0.7195	0.1723	19.33
- ,	0.5	1.7601	0.505	0.5281	51.12
	1.0	1.0705	0.3072	0.7858	71.89

^{*}Treatment time = 24 h

^{**}Treatment time = 48 h

The data in Table 1 show that the amount of dissolved collagen increases when HCl concentration exceeds 0.5 N. Thus, we recommend using a 0.5 N HCl solution for dissolving the fossil bone nonorganic component, and 1.0 N HCl for well-preserved, compact bones and teeth. L. D. Sulergitskiy (personal communication) established that collagen yield increases considerably if HCl treatment is performed at low temperature (in a refrigerator, for example). Table 1 also shows that collagen solubility increases greatly when NaOH concentration is >0.1 N. For this reason, humic matter should be extracted with a ≤0.1N NaOH solution (Arslanov and Gromova 1970).

We assumed that the best conditions for humus extraction can be achieved by purifying collagen via extraction of soluble humus impurities with 0.1 N NaOH at room temperature, then precipitating nearly insoluble impurities from the collagen extraction using Longin's method (Arslanov, Svezhentsev and Markov 1981). As a control, we dated a large number of ancient fossil bone samples, using these two methods and our own.

The bones were mechanically cleaned and washed, then pulverized and treated at low temperature $(4-6^{\circ}\text{C})$ by 2-3 fresh solutions of 0.5-1.0 N HCl for a few days (depending on preservation condition) until mineral components dissolved completely. We washed the collagen obtained in distilled water until no Ca^{2+} was detectable. We then treated the collagen with 0.1 N NaOH at room temperature for 24 h, and washed it again in distilled water until neutral. We treated the collagen with a weak HCl solution (pH = 3) at $80^{\circ}-90^{\circ}\text{C}$ for 6-8 h. Finally, we separated the humic acid residue from the gelatin solution by centrifugation, and the solution was evaporated. Benzene was synthesized from the dried gelatin by burning in a "bomb" or by dry pyrolysis, using the standard methods (Arslanov, Svezhentsev and Markov 1981). Table 2 shows the results.

DISCUSSION AND CONCLUSIONS

The data in Table 2 show that the oldest measured age is achieved by extracting the collagen fraction by the method that we recommend here. The fraction extracted using Longin's method has the youngest age. The age discrepancy of these fractions is caused by incomplete removal of younger humic substances. Thus, the experimental results observed earlier, as well as new data presented here, demonstrate incomplete removal of humic substances by a hot, weak HCl solution (Longin's method) or by a 0.1 N NaOH solution (Arslanov, Svezhentsev and Markov 1981). This illustrates that a single method, especially Longin's, should not be used alone in dating old samples. Recently, Gurfinkel (1987) compared different methods of collagen purification and concluded that the acid-alkali method or the two-stage combined method is preferable to Longin's method. Our current procedure (Arslanov, Svezhentsev and Markov 1981) allows us to remove separate soluble and nearly insoluble forms of humic matter, and to obtain more reliable ages. Hakansson (1976) first used the two-stage combined method for ¹⁴C dating fossil bones, but did not provide any experimental data.

We compared the combined treatment method to the technique of extracting collagen with EDTA (Olsson *et al.* 1974). Table 3 presents ¹⁴C dates measured from samples prepared using each method (Arslanov and Svezhentsev 1983). The data clearly indicate that samples treated by the combined method yield older ages.

From these results, we cannot recommend the EDTA method for dating ancient bones; further, the EDTA procedure is laborious. We recommend the combined method with which we can remove more completely exogenous carbon contamination and obtain more reliable dates. The method may be used for dating by conventional β-counting or by accelerator mass spectrometry.

TABLE 2. ¹⁴C Dates on Bones Obtained by Different Chemical Treatments

			¹⁴ C ages (yr BP)		14C some for
Site	Material	Longin method	HCl – NaOH method	Combination method	charcoal or other material
Golubaja I, Krasnojarsk region	Bone	12,380 ± 140 (LE-1101A)	$12,900 \pm 150$ (LE-1101B)	13,650 ± 180 (LE-1101C)	13,050 ± 120 (T.F-1101D) charred bone
Gagarino, Lipeckanian region	Mammoth tooth	$(17,930 \pm 100)$ (LE-1432A)	$20,150 \pm 300$ (LE-1432B)	$20,620 \pm 300$ (LE-1432C)	21,800 ± 300 (GIN-1872) charred bone
Kostenki XI, Voronezh region, Layer la	Bone	$14,610 \pm 120$ (LE-1637)	$(16,040 \pm 120)$ (LE-1704A)	$(17,310 \pm 200)$	19,900 ± 350 (GrN-2532) charred bone
Kostenki XII, Layer Ia	Bone	$28,700 \pm 400$ (LE-1428A)	$30,240 \pm 400$ (LE-1428B)	$31,150 \pm 400$ (LE-1428C)	$32,700 \pm 700$ (GIN-7758) charcoal
Kostenki XXI, lower Layer III	Bone	$19,100 \pm 150$ (LE-1437A)	$20,250 \pm 100$ (LE-1437B)	$(22,900 \pm 150)$ (LE-1437C)	$22,270 \pm 150$ (GrN-7363) charcoal
Jidinovo, northern bank of Sudost River, Brjansk region, Dugout II	Bone	$15,790 \pm 320$ (LE-3301)		(LE-3302A)	17,800 ± 810 (LE-3302B) charred bone
Anetovka II, bank of Bakarsha River, Nikolaevsk region	Bone	$18,000 \pm 150$ (LE-2424A)	$18,260 \pm 1650$ (LE-2424B)	19, 170 ± 120 (LE-2424C)	
Sagaidak I, 1st terrace of South Bug River, Nikolaevsk region	Mammoth tooth	$20,300 \pm 200$ (LE-1602A)	\ 	$21,240 \pm 200$ (LE-1602B)	
Ui-I, Tuva Autonomous Republic	Bone	$16,760 \pm 120$ (LE-3358)	$17,520 \pm 130$ (LE-3359A)	$19,280 \pm 200$ (LE-3359B)	
Maininskaya, Layer 4, Tuva Autonomous Republic	Bone	$12,910 \pm 100$ (LE-2133A)	$12,980 \pm 130$ (LE-2133B)	$13,690 \pm 390$ (LE-2133C)	
Kurtak-4, upper layer, depth 5 m, terrace, Yenisei Krasnojarsk region	Bone	23,470 ± 200 (LE-3351)	$23,800 \pm 900$ (LE-3357)	24,890 ± 670 (LE-2833)	

TABLE 3. Comparison of Methods for Collagen Dating of Fossil Bones

		¹⁴ C ages (BP)		
Site	Material	Combined method	EDTA method	
Kostenki XXI, depth	Bone	22,900 ± 150	19,050 ± 130	
1.55-1.73 m		(LE-1437C)	(LE-1437D)	
Dvuglaska Cave, Layer 6,	Bone	>40,210	28,030 ± 290	
Krasnojarsk region		(LE-2346)	(LE-2347)	
Maininskaya Layer 3, Tuva	Bone	14,070 ± 150	12,330 ± 150	
Autonomous Republic		(LE-2149A)	(LE-2149B)	
Kurtak-4 upper layer, depth 5m, terrace of Yenisei River, Krasnojarsk region	Bone	24,890 ± 670 (LE-2833)	24,170 ± 230 (LE-3351)	
Junovo, near Junovo village, terrace of Yenisei River, Krasnojarsk region	Bone	36,500 ± 500 (LE-1435A)	35,000 ± 500 (LE-1435B)	
Leski zone, South Bug River,	Mammoth tooth	23,770 ± 750	19,200 ± 200	
Nikolaevsk region		(LE-2946A)	(LE-2946B)	

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