Terahertz Rotational Spectroscopy

T. F. Giesen¹, S. Brünken², M. Caris¹, P. Neubauer-Guenther¹, U. Fuchs¹, G. W. Fuchs³ and F. Lewen¹

¹I. Physikalisches Institut, Universität zu Köln, Zülpicher Str. 77, 50937 Cologne, Germany

²Harvard-Smithsonian Center for Astrophysics, 60 Garden Street, Cambridge, MA 02138, USA

³Raymond and Beverly Sackler Laboratory for Astrophysics, Leiden Observatory, Leiden University, Postbus 9513, 2300 RA, Leiden, The Netherlands

Abstract. The detection of interstellar molecules relies on the precise knowledge of spectral line positions from laboratory measurements. Technical developments of recent years have led to an extension of the accessible spectral range towards shorter wavelengths. New telescopes like SOFIA, the HIFI instrument aboard the Herschel satellite, and ALMA will be used for astrophysical observations in the terahertz region. The Cologne group has developed precise spectrometers to study molecules of astrophysical importance under laboratory conditions and to obtain characteristic spectra for their possible detection in space. We present recent results on light hydrides, carbon-chain molecules and more complex species.

Keywords. ISM: molecules — methods: laboratory

1. Introduction

Radioastronomy has been quite successful over the past decades in detecting new molecules in space. So far, more than 130 species have been clearly identified by means of characteristic spectral lines, and the number of new detections grows on average by four per year. While most of the detections were made at centimeter- and millimeterwavelengths, new receiver technologies are opening up the spectral range towards shorter wavelengths and leading into the sub-millimeter wavelength region, also known as the terahertz domain. A number of new telescopes, such as the Atacama Pathfinder Experiment, APEX, the Stratospheric Observatory for Infrared Astronomy, SOFIA, and the HIFI instrument aboard the Herschel satellite will explore the terahertz domain and will certainly lead to the detection of new interstellar molecular species. This identification of new molecules relies on precise laboratory spectra. It is, thus, a challenging task for laboratory groups world wide to provide precise spectral data of molecules which might be of astrophysical importance.

Over the last two decades, the Cologne group has developed precise spectrometers to cover the spectral range from 50 GHz to 2.7 THz. In this paper we emphasize particular investigations of molecules that have characteristic spectra in the terahertz region, and which can be assigned to three important classes:

- (1) pure rotational spectra of light hydrides,
- (2) low ro-vibrational bending modes of carbon-chain molecules,
- (3) spectra of large molecules with internal motions.

We will discuss these three aspects of terahertz spectroscopy by giving examples of molecules, which have recently been studied at the Cologne laboratories, and which are of particular interest for astronomy. In § 2 we present our recent results on the methylene radical, CH_2 , a light hydride having pure rotational transitions from the lowest energy



Figure 1. Frequency coverage of the Cologne spectrometers and the observatories SOFIA, Herschel and APEX.

levels at 2–3 THz. Ro-vibrational bending mode transitions of many molecules occur predominantly in the mid-infrared region, whereas carbon chains possess unusually low frequency bending vibrations. For example, the ν_2 mode of C₃ has been found at 2 THz and high-level *ab initio* calculations predict the lowest vibrational modes of longer pure carbon chains, such as C₄, C₅,...,C₁₀ to lie in the range of 1–5 THz. In § 3 we discuss the spectroscopy of carbon-chain molecules and present recent results on C₃H. § 4 deals with the spectral complexity of larger molecules and recent results on ethyl methyl ether, C₂H₅OCH₃. § 5 contains some general remarks on future spectroscopic investigations in the THz region and on the necessity of a data base for molecular spectroscopy.

The Cologne THz-spectrometers use backward wave oscillators (BWOs) as radiation and fundamental frequency sources to cover the frequency range from 50 GHz to 1.1 THz. Two different approaches are followed to extend the accessible frequency range beyond this point. First, the BWO output radiation is mixed with the radiation of a fixed frequency far infrared laser to generate sideband radiation between 1.75–2.01 THz. Secondly, to cover the gap between 1–2 THz and, furthermore, to access frequencies up to 2.7 THz, efficient superlattice multiplier devices (Scheuerer *et al.* 2003), pumped by the radiation of BWOs at 100–300 GHz are used. Figure 1 shows the spectral coverage of our spectrometers and of some present and future telescope receivers.

High spectral accuracy, $\Delta f/f = 10^{-11}$, is obtained by phase-locking the BWO to a Rubidium reference, and spectra can be recorded with Doppler- and partially sub-Doppler limited resolution. Detailed descriptions of the spectrometers can be found elsewhere (Winnewisser *et al.* 1994). The spectrometers have been combined with several devices for the production of unstable and highly-reactive molecules. These molecular sources include discharge cells, radio plasma techniques, and supersonic jets.

2. Light Hydrides – The Methylene Radical, CH₂

Large rotational constants are characteristic for light hydrides, defined here as molecules of the form XH_n (n = 1, 2). Consequently, even their ground state rotational transitions appear at comparatively high frequencies. At Cologne, the rotational spectra of a multitude of light hydride species have been measured during the last years by employing the Cologne terahertz spectrometers. Among these are the water molecule, H₂O, (Brünken *et al.* 2005), the NH radical (Lewen *et al.* 2004), the amidogen radical NH₂ (Gendriesch *et al.* 2001), and the phosphidogen radical PH₂ (Margulès *et al.* 2002). The methylene radical, CH₂, is an important molecule for both spectroscopists and astrochemists. It is expected to be abundant in both dense and diffuse interstellar clouds and to be presumably produced quite early in the sequence of ion-molecule reactions, which makes it a test molecule for chemical models (Lee *et al.* 1996). Due to its lightness, the lowest levels are already high in energy and are widely spaced in energy. Thus transitions which probe the lowest rotational states occur in the terahertz spectral region. In its electronic ³B₁ ground state it has a bent, symmetric structure with an H-C-H equilibrium bond angle of 133.9308(21) degrees (Jensen & Bunker 1988) and a low barrier to linearity of less than 2000 cm⁻¹. A very shallow potential energy curve causes its floppiness, which in turn leads to strong centrifugal distortion effects. The approach of a standard Watson-type A-reduced Hamiltonian fails in the case of light and floppy molecules and leads to large uncertainties for the prediction of transition frequencies, especially at higher rotational quantum numbers.

CH₂ is an asymmetric top molecule with a permanent dipole moment of 0.57 Debye (Bunker & Langhoff 1983) pointing along the *b* molecular axis, which is the bisecting line of the H-C-H angle. Although CH₂ is a simple triatomic molecule, its spectrum is fairly complicated due to fine and hyperfine interactions. The electron spin in a ${}^{3}B_{1}$ ground state (S = 1) causes a splitting of each rotational energy level into three fine structure components. Due to the presence of two protons, CH₂ has both an *ortho* (total nuclear spin, I = 1) and a *para* (I = 0) form. In the case of *ortho*-CH₂, the nuclear spin causes a further splitting of each fine structure rotational level into three hyperfine levels.

The first laboratory measurements on CH₂ were performed by Herzberg & Shoosmith (1959), which led to a clear identification of the ${}^{3}B_{1}$ ground state and later to the determination of its structure. In the beginning of the 1980s, Sears et al. used laser magnetic resonance spectroscopy to record the first ro-vibrational spectra in the midinfrared (Sears et al. 1981, 1982) and pure rotational spectra of the vibrational ground state in the far infrared region (Sears et al. 1982a). These data were used by Lovas et al. (1983) to perform the first measurements on the $4_{04} - 3_{13}$ pure rotational transition at \sim 70 GHz with microwave accuracy. Hollis *et al.* (1989) used the 70 GHz transition to start several attempts for an interstellar detection of CH₂ towards the Orion nebula. The tentative detection was later confirmed by the same authors using a receiver of improved sensitivity (Hollis et al. 1995). Ozeki & Saito (1995) measured two rotational transitions of $para-CH_2$, which correspond to excitation temperatures as high as 135 K and 310 K respectively. A search for CH_2 in the cold interstellar medium requires precise data of transitions, where the lowest energy levels are involved. Because of its floppiness these data can not be obtained to high accuracy from predictions based on a standard Watson Hamilton approach. To overcome these constraints, we started extensive laboratory studies on CH_2 in 2003 with the clear intention of extending the sparse data set towards terahertz frequencies and supporting a decisive astrophysical search for CH_2 in cold interstellar clouds.

We produced the methylene radical from ketene, CH₂CO, which was disintegrated in a DC-discharge cell cooled to a temperature of roughly 170 K to populate the lower-energy levels of CH₂. A 4 kHz Zeeman modulation was employed for phase sensitive signal detection, making use of the permanent magnetic moment of methylene. We used the Cologne THz spectrometer for measuring the hyperfine components of the $N_{Ka,Kc} = 1_{1,1} - 2_{0,2}$ transition of ortho-CH₂ centered at 943 GHz (Michael et al. 2003) and recorded two energetically low-lying transitions near 2 THz (Brünken et al. 2004) by operating the **Co**logne Sideband Spectrometer for Terahertz Applications (COSSTA; Lewen et al. 1997). One of those transitions, the $N_{KaKc} = 1_{1,0} - 1_{0,1}$ at 1.915 THz stems from the lowest para energy level and is best suited for a detection of para-CH₂ in a cold interstellar

environment. To fit the data to an effective Hamiltonian, we used a new approach introduced by Pickett et al. (2005), which is based on an expansion of the angular momentum operators in terms of an Euler series. All our data and all available precise data from the literature were included in the fit to give the most reliable set of molecular constants available (Brünken et al. 2005a). Based on this analysis, transition frequencies from the lowest energy levels of CH_2 were calculated and used by Polehampton *et al.* (2005) to search for CH_2 in the cold interstellar medium. For this purpose, the authors analyzed data obtained with the Long Wavelength Spectrometer (LWS; Clegg et al. 1996) aboard the Infrared Space Observatory (ISO). Several fine structure components of three rotational transitions of CH_2 were detected in cold molecular clouds in the line of sight towards the galactic center source Sagittarius B2 (Sgr B2). These transitions involve the three lowest rotational energy levels of methylene. The detections allowed the determination of the CH_2 column density and the estimation of the *ortho/para* ratio in the observed regions. Furthermore, the simultaneous observation of CH and CH₂ absorption lines gave insight into the underlying chemistry by comparison of the measured abundance ratio to that derived from different theoretical models. These astrophysical investigations were triggered by and based upon state-of-the-art laboratory measurements and their subsequent analysis.

3. Low Bending Modes of Carbon-Chain Molecules

Carbon is the fourth most abundant element in space. A large fraction of the more than 130 different interstellar molecules contain at least one carbon atom. Due to its unique ability to form linear chains, rings and branched, three dimensional structures, carbon is the most important element of a complex interstellar chemistry. It is conspicuous that many interstellar detected molecules are highly unsaturated linear carbon chains. Chemical models also predict *pure* carbon chains to be quite abundant in interstellar clouds and in the environment of late type carbon stars; nevertheless, only C_3 and C_5 have been undoubtedly detected so far. In the 19th century, Huggins (1881) discovered a spectral feature in the optical spectrum of comet b 1881 which was later assigned to C_3 by Douglas (1951). Based on ro-vibrational transitions in the infrared Hinkle *et al.* (1988) detected C_3 in the shell of the carbon star IRC+10216. Recent detections by rotationally resolved spectra in the visible range were published by Maier *et al.* (2001), who found C_3 in three diffuse interstellar clouds towards reddened stars and by Adamkovics *et al.* (2003) in translucent sight lines. The only secure detection of C_5 was reported by Bernath *et al.* (1989) towards IRC+10216 based on infrared spectra.

The main reason for the almost complete absence of pure carbon chains from the list of interstellar molecules is the missing of a permanent dipole moment, which would allow their detection by pure rotational transitions in the millimeter-wave region. However, pure carbon chains are predicted to have low-lying bending vibrations in the terahertz region. Precise knowledge of these transition frequencies would enable astronomers to search for longer linear carbon chains in cold interstellar clouds. Ab initio calculations on the structure and the vibrational frequencies were performed by e.g. Martin & Taylor (1996) and Botschwina (2005) by means of coupled cluster calculations. The bending mode vibrational dipole moment is on the order of 0.2–0.3 Debye for most of the C_n chains with n = 3-14.

 C_3 is the only pure carbon-chain molecule for which measurements of the lowest bending mode transitions have been performed, using tunable far-infrared laser sideband spectroscopy at 2 THz (Schmuttenmaer *et al.* 1990; Gendriesch *et al.* 2003). The measurements by Schmuttenmaer *et al.* led to the detection of C_3 in the source Sgr B2,



Figure 2. C_3 absorption line toward Sgr B2 using the KAO. The peak was assigned to the R(2) transition of the ν_2 bending mode.

which was the first detection of a pure carbon molecule in an interstellar cloud at all (Van Orden et al. 1994 and Giesen et al. 2001; see Fig. 2). This one-line detection was later confirmed by Cernicharo et al. (2000), using the long wavelength Fabry-Perot spectrometer (LWS) aboard the ISO satellite to detect C_3 in Sgr B2 and IRC+10216. The terahertz region will be discovered by a new generation of telescopes, such as SOFIA and the HIFI instrument aboard the Herschel satellite. Due to the lack of accurate laboratory data, it is inevitable to perform high-resolution measurements on the bending modes in this spectral region. For this purpose we have combined our supersonic jet molecular sources with a new Terahertz-Spectrometer. Figure 3 shows the experimental setup. The Supersonic Jet Spectrometer for Terahertz Applications (SuJeSTA) is an instrument that allows high-resolution mm- and submm-wave spectroscopy of unstable, highly-reactive molecules, such as radicals and ions at low rotational temperatures of a few Kelvin. Highly-reactive molecules are produced in either an electrical discharge, placed in the throat of the source, or by laser ablation of a graphite target. Both techniques are well established and have been described in various papers (see e.g. Linnartz et al. 2000; Giesen et al. 1994).

A variety of molecules can be produced by an appropriate mixture of gases entering the discharge source. The jet is probed by the radiation of tunable BWOs operating between 300 and 1200 GHz and very recently in combination with superlattice devices up to 2.7 THz. The spectrometer can be operated in two modes, a frequencystabilized mode for highly-precise measurements over a small range of up to 500 MHz (typically 10–20 MHz), or in a free-running fast-scanning mode, which covers a large frequency range of several GHz but with less accuracy. The fast-scanning mode is used for searching the absorption lines of less known, or even completely unknown species, whereas the frequency-stabilized mode is used for highly-precise measurements (10 kHz accuracy) on selected, identified lines and for long integration times of weak signals. The technique of frequency stabilization of BWOs is well established at the Cologne



Figure 3. Experimental setup of the Cologne Supersonic Jet Spectrometer for Terahertz Applications (SuJeSTA). Backward Wave Oscillators (BWOs) are used as radiation sources. The signal is detected by a liquid helium cooled InSb-hot electron bolometer. A pulsed discharge slit nozzle produces C_3H radicals in a supersonic expansion of a C_2H_2/CO gas mixture diluted in He or Ar at 1–6 bar backing pressure. The probing microwave beam intersects the molecular jet six times in a multi-pass optical array, which leads to an overall absorption pathlength of approximately 30 cm.

laboratories (Winnewiser *et al.* 1994) and is used routinely in all of our measurements to achieve most reliable spectroscopic data for both laboratory and astrophysical applications.

Among a number of carbon-containing molecules detected in space, C_3H is considered a key molecule of carbon-chain growth in dark clouds. There are two different isomers of C_3H known, the linear $(l-C_3H)$ and the cyclic $(c-C_3H)$ forms, both of which have been identified in space by means of radioastronomy (Turner *et al.* 2000). The *ab initio* geometry for $l-C_3H$ in the ground state has been calculated to be slightly bent with a rather small barrier to linearity of 100–200 cm⁻¹. The linear C_3H radical was detected by Thaddeus *et al.* (1985) towards IRC+10216 and TMC-1; laboratory spectra in the ground vibrational state were observed and analyzed by Gottlieb *et al.* (1986). The first information on rotational transitions of vibrationally excited states was obtained by Yamamoto *et al.* (1990). Although they deduced the existence of an extremely low lying $^{2}\Sigma$ bending state from their measurements, they were unable to directly measure transitions from the two $^{2}\Pi$ electronic ground state levels into the lowest $^{2}\Sigma$ bending state.

We have measured high-resolution spectra of the ²II vibrational ground state of C₃H and rotationally resolved transitions of the ²II –² Σ bending vibration. We used SuJeSTA to perform measurements in the frequency range between 370 and 600 GHz to extend the data set of Yamamoto *et al.* (1990) towards higher frequencies. A Hamiltonian for ²II molecules in a Hund's case (*a*) basis set has been used for a least squares fit analysis of all new data, including the measurements up to 350 GHz published by Yamamoto *et al.* (1990). We obtained improved molecular parameters and were able to determine the exact position of the low-lying ² Σ band for the first time.



Figure 4. Energy level diagram for an asymmetric top molecule with internal rotation for the energy levels J = 37 with $K_a = 4, 5$. Asymmetric splitting, as well as splitting due to internal rotation, is depicted.

4. Complex Molecules – The Case of Ethyl Methyl Ether

During the process of high-mass star formation in hot-core regions of interstellar clouds temperatures can rise to 100-300 K and subsequently lead to evaporation of icy mantles from interstellar dust grains (Millar et al. 1998). The mantles are thought to contain alcohols such as methanol (CH_3OH) and ethanol (C_2H_5OH), which have been produced via surface reactions (Charnley et al. 1995). In the gas phase, these alcohols can react with protonated alcohols to form ethers. Dimethyl ether (CH_3OCH_3) is the simplest representative of this class and has been found in hot cores (Ikeda et al. 2001). It is thus very likely to find larger ethers, such as ethyl methyl ether $(CH_3CH_2OCH_3)$, in hot-core regions. Charnley et al. (2001) observed a single line in W51 e1/e2 and Orion KL at 160.1 GHz and another at 79.6 GHz in Sgr B2(N), which they assigned to trans-ethyl methyl ether. Since warm interstellar regions possess a rich interstellar chemistry, they have fairly dense spectra, and consequently these single line detections remained tentative. An unambiguous confirmation was highly desirable. We therefore started a laboratory investigation of ethyl methyl ether, using the FASSST spectrometer at Columbus, Ohio and the Cologne Terahertz spectrometer to record precise spectra from 55 GHz to 350 GHz. Ethyl methyl ether is an asymmetric top molecule with a strong b-type and a weaker a-type spectrum. Figure 4 shows the energy levels of the $J_K = 37_4$ and $J_K = 37_5$ rotational states. Asymmetry splitting lifts the degeneracy of the rotational levels, and the energy level diagram is further complicated by the internal rotation of the two methyl groups, which causes each level to split into five sub-levels.

From the analysis of more than 1000 individual lines including the lower-frequency transitions up to 40 GHz, which had been published earlier by Hayashi & Kuwada (1975), precise molecular parameter were derived to give reliable line frequency predictions up to 400 GHz (Fuchs *et al.* 2003). In 2002 and 2003, we used the IRAM 30-m telescope at Pico Veleta, Spain, and the SEST 15m telescope at La Silla, Chile, to search for ethyl methyl ether in several hot-core regions (Fuchs *et al.* 2005). The spectra contain lines from other organic molecules; most prominent are methanol lines, which overlap with many line positions of ethyl methyl ether. For all sources, we were able to find a set of ethyl methyl ether transition frequencies that are not blended by any other constituent. We clearly detected ethyl methyl ether towards W51 e2 and estimated a column density of 2×10^{14} cm⁻² at $T_{rot} = 70$ K. Although these measurements confirm the tentative detection towards W51 e2 published by Charnley *et al.* (2001), we cannot confirm their tentative detection of ethyl methyl ether in Orion KL. In W51 e2 we also determined the relative abundance of ethers.

5. Conclusions

Ground-based observations in the terahertz region are hampered or completely blocked by the earth atmosphere. With the advent of new telescopes such as SOFIA and the Herschel/HIFI instrument, this important spectral window will be opened up for astronomical observations. We are thus expecting a wealth of line detections that carry information on well-known but also on completely unknown interstellar molecules. These new measurements will provide a deeper insight into physical conditions of astronomical sources and their molecular inventory. Accomplishing laboratory measurements in the terahertz region will assist the identification of spectral features and will play a key role in the detection of new molecules in space. The development of highly-precise terahertz spectrometers is thus of essential importance. New multiplier techniques based on superlattice devices will expand the spectral range to several THz. In addition, monochromatic radiation sources such as quantum cascade lasers, which are commercially available for the mid-infrared region, will reach to the far-infrared region in the near future.

The Cologne spectrometers are well suited for the investigation of highly-reactive molecules, radicals and molecular ions in the terahertz region. Here we have presented three categories of molecules, which are of astrophysical importance. Light hydrides are thought to occur early in the formation of interstellar molecules and are appropriate candidates to test chemical models of ion-molecule and grain-gas phase reactions. Especially for di-hydrogenated species in cold environments the *ortho/para* ratio can deviate from the one expected from spin statistics and gives clues about the molecular formation.

Carbon-chain molecules are abundant in cold interstellar clouds and have been found in the shells of late type carbon stars. It is a challenging task to find *pure* carbon chains in theses objects. Although they have been predicted from chemical networks, they have so far eluded astronomical detections owing to the lack of laboratory data. We will use the Cologne spectrometers to study the terahertz spectra of pure carbon chains in the laboratory and will give precise data for their detection in space. Not only pure carbon chains, but all centro-symmetric chain molecules such as HC_nH , OC_nO , and NC_nN , (n = 3, 4, ...) have no permanent electrical dipole and thus no pure rotational spectrum. However, they have their lowest bending modes in the terahertz region, as has been predicted from *ab initio* calculations (Botschwina 2005).

Interstellar carbon chemistry bears a variety of complex molecules, which usually have quite complicated spectra spread over a wide frequency range. Laboratory data are rather sparse, but highly demanded for a more complete inventory of complex molecules in space. The Cologne spectrometers are well suited to record spectra over a wide frequency range from 50 GHz to 2.7 THz and will be used to investigate molecules such as dimethyl ether, ethyl formate, and ethyl cyanide at high frequencies.

Laboratory spectroscopists and astrophysicists use different languages when they talk about molecules. While spectroscopists use parameters and constants to describe molecules, astrophysicists prefer frequencies and line positions. The Cologne spectroscopy group has therefore created the Cologne Data Base for Molecular Spectroscopy (CDMS), as a link between laboratory spectrocopy and astronomy. The CDMS, which contains line catalogs of more than 300 species, uses the same format as the JPL catalog. Free access is granted via internet http://www.cdms.de.

Acknowledgements

The present investigations were supported by the Deutsche Forschungsgemeinschaft under the contracts DFG-SFB 494 and LEA *HiRes*.

References

Adamkovics, M., Blake, G.A., & McCall, B.J. 2003, Ap. J. 595, 235

- Bernath, P.F., Hinkle, K.W., & Keady, J.J. 1989, Science 244, 562
- Botschwina, P. 2005, Mol. Phys. 103, 1441 and references therein
- Brünken, S., Michael, E.A. Lewen, F., Giesen, T., Ozeki, H., Winnewisser, G., Jensen, P., & Herbst, E. 2004, Can. J. Chem. 82, 676
- Brünken, S. 2005, Ph.D. thesis, High Resolution Terahertz Spetctroscopy on Small Molecules of Astrophysical Importance (Göttingen: Cuvillier Verlag)
- Brünken, S., Müller, H.S.P., Lewen, F., & Giesen, T.F. 2005a, J. Chem. Phys. 123, 164315
- Bunker, P.R. & Langhoff, S.R. 1983, J. Mol. Spect. 102, 204
- Cernicharo, J., Goicoechea, J.R., & Caux, E. 2000, Ap. J. 534, L199
- Charnley, S.B., Kress, M.E., Tielens, A.G.G.M., & Millar, T.J. 1995, Ap. J. 448, 232
- Charnley, S.B., Ehrenfreund, P., & Kuan, Y.-J. 2001, Spectrochim. Acta A 57, 685
- Clegg, P.E., Ade, P.A.R., Armand, C. et al. 1996, A&A 315, L38
- Douglas, A.E. 1951, Ap. J. 94, 466
- Fuchs, G.W., Fuchs, U., Giesen, T.F., & Wyrowski, F. 2005, A&A 444, 521
- Fuchs, U., Winnewisser, G., Groner, P., De Lucia, F. C., & Herbst, E. 2003, Ap. J. Suppl. 144, 277
- Gendriesch, R., Lewen, F., Winnewisser, G., & Müller, H.S.P. 2001, J. Mol. Struct. 599, 293
- Gendriesch, R., Pehl, K., Giesen, T., Winnewisser, G., & Lewen, F. 2003, Z. Naturforsch. 58a, 129
- Giesen, T.F., Van Orden, A., Hwang, H.J., Fellers, R.S., Provencal, R.A., & Saykally, R.J. 1994, Science 265, 756
- Giesen, T.F., Van Orden, A.O., Cruzan, J.D., Provencal, R.A., Saykally, R.J., Gendriesch, R., Lewen, F., & Winnewisser, G. 2001, Ap. J. 551, L181
- Gottlieb, C.A., Gottlieb, E.W., Thaddeus, P., & Vrtilek, J.M. 1986, Ap. J. 303, 446
- Hayashi, M. & Kuwada, K. 1975, J. Mol. Struc. 28, 147
- Herzberg, G. & Shoosmith, J. 1959, Nature 183, 1801
- Hinkle, K.W., Keady, J.J., & Bernath, P.F. 1988, Science 241, 1319
- Hollis, J.M., Jewell, P.R., & Lovas, F.J. 1989, Ap. J. 346, 794
- Hollis, J.M., Jewell, P.R., & Lovas, F.J. 1995, Ap. J. 438, 259
- Huggins, W. 1881, Proc. R. Soc. London 33, 1
- Ikeda, M., Ohishi, M., Nummelin, A., Dickens, J.E., Bergman, P., Hjalmarson, A., & Irvine, W.M. 2001, Ap. J. 560, 792
- Jensen, P. & Bunker, P.R. 1988, J. Chem. Phys. 89, 1327
- Lee, H.-H., Bettens, R.P.A., & Herbst, E. 1996, A&A 119, 111

- Lewen, F., Michael, E., Gendriesch, R., Stutzki, J., & Winnewisser, G. 1997, J. Mol. Spect. 183, 207
- Lewen, F., Brünken, S., Winnewisser, G., Simeckova, M., & Urban, S. 2004, J. Mol. Spect. 226, 113
- Linnartz, H., Vaizert, O., Motylewski, T., & Maier, J.P. 2000, J. Chem. Phys. 112, 9777
- Lovas, F.J., Suenram, R.D., & Evenson, K.M. 1983, Ap. J. 267, L131
- Maier, J.P., Lakin, N.M., Walker, G.A.H., & Bohlenender, D.A. 2001, Ap. J. 553, 267
- Margulès, L., Herbst, E., Ahrens, V., Lewen, F., Winnewisser, G., & Müller, H.S.P. 2002, J. Mol. Spect. 211, 211
- Martin, J.M.L. & Taylor, P.R. 1996, J. Phys. Chem. 100, 6047
- Michael, E.A., Lewen, F., Winnewisser, G., Ozeki, H., Habara, H., & Herbst, E. 2003, Ap. J. 596, 1356
- Millar, T.J. & Hatchell, J. 1998, Faraday Discuss. 109, 15
- Ozeki, H. & Saito, S. 1995, Ap. J. 451, L97
- Pickett, H.M., Pearson, J.C., & Miller, C.E. 2005, J. Mol. Spect. 233, 174
- Polehampton, E., Menten, K., Brünken, S., Winnewisser, G., & Baluteau, J.-P. 2005, A&A 431, 203
- Scheuerer, R., Haeussler, M., Renk, K.F., Schomburg, E., Koschurinov, Y.I., Pavelev, D.G., Maleev, N., Ustinov, V., & Zhukov, A. 2003, Appl. Phys. Lett. 82(17), 2826
- Schmuttenmaer, C.A., Cohen, R.C., Pugliano, N., Heath, J.R., Cooksy, A.L., Busarow, K.L., & Saykally, R.J. 1990, Science 249, 897
- Sears, T.J., Bunker, P.R., & McKellar, A.R.W. 1981, J. Chem. Phys. 75, 4731
- Sears, T.J., Bunker, P.R., & McKellar, A.R.W., 1982, J. Chem. Phys. 77, 5363
- Sears, T.J., Bunker, P.R., McKellar, A.R.W., Evenson, K.M., Jennings, D.A., & Brown, J.M. 1982a, J. Chem. Phys. 77, 5348
- Thaddeus, P., Gottlieb, C.A., Hjalmarson, A., Johansson, L.E.B., Irvine, W.M., Friberg, P., & Linke, R.A. 1985, Ap. J. 294, L49
- Turner, B.E., Herbst, E., & Terzieva, R. 2000, Ap. J. Suppl. 126, 427
- Van Orden, A., Cruzan, J.D., Provencal, R.A., Giesen, T.F., Saykally, R.J., Boreiko, R.T. Boreiko, & Betz, A.L. 1994, Proceedings of the 1994 Kuiper Airborne Astronomy Symposium, NASA-Ames Research Center
- Yamamoto, S., Saito, S., Suzuki, H., Deguchi, S., Kaifu, N., Ishikawa, S., & Ohishi, M. 1990, Ap. J. 348, 363
- Winnewisser, G., Krupnov, A.F., Tretyakov, M.Y., Liedtke, M., Lewen, F., Saleck, A.H., Schieder, R., Shkaev, A.P., & Volokhov, S.V. 1994, J. Mol. Spec. 165, 294

Discussion

GUÉLIN: Did you study the bending modes of C_4H in the same way as you studied those of C_3H ? Rotational transitions within vibrationally excited states of C_4H are observered in IRC+10216 and the lines are surprisingly strong. It would be interesting to learn how difficult it is to excite these states.

GIESEN: C_3H was the first molecule we studied with the new SuJeSTA instrument. C_4H indeed would be a very interesting molecule to study in the lowest bending mode, although the predicted frequency of 226 cm⁻¹ or 7 THz is too high for our spectrometers.

HERBST: Can you help identify ethyl methyl ether in space by its internal rotation splittings?

GIESEN: In W51 e2 and Orion KL the internal dynamics of the source cause a line broadening that is larger than the internal rotation splittings of ethyl methyl ether. But in more quiescent sources, or at higher spatial resolution, the splittings would be partially resolved and could be used as characteristic patterns to identify ethyl methyl ether lines.